



Heat Thermal Battery Storage



We have further developed the worlds most efficient thermal storage device.
It can hold very high temperatures in small spaces for long periods of time.
We can, we call it the Hot Box.

STORE RENEWABLE ENERGY

- Solar PV
- Solar Thermal
- Wind
- Hydro
- Geothermal
- Biogas
- Natural Gas

Heat Thermal

Battery Storage

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The CHP Orc Turbine By Secure Supplies and the Injected Liquid Heat Fischer engine runs on Pressure Expansion. No fuel is necessary to power it, but you do need a heat source.

The Hot Box act as a capacitor for Secure Supplies Heat and absorbs heat through electric

Cal-rods or by circulating a heat transfer liquid.

The Fischer Generator takes energy out of the air for free at a persons home , Business or Vehicle.

Global Distribution

Quality Management System:



ISO9001-2015

GJB9001B-2009 (MIL)

Secure Supplies Product/s Meet Customers Demands.

Authorized Representative



Website

<https://www.secure.supplies/heat-thermal-battery-storage>





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Secure Supplies Low Pressure Heat Pump Systems can charge the Hot Box in their location with their Fischer Generator at night while they are sleeping.

A special Hot Box model was made for cars as well as other heat storage models. Through special state of the art insulation, the very high temperature heat could remain in the Hot Box all day (or even all week long).

By the way, there is another type of engine cycle that BWT makes that can run your engines or turbine run an electric generator without exhaust or the need for a condenser.

It is called the ORC Turbine or another is Sterling Engine Cycle.

It runs just on heat. All you need is a heat source. Secure Supplies has Master the art of pressure with out heat.

With the Hot Box, your vehicles or power plants could also be powered with this engine type.

What is a thermal battery?

Thermal mass of any kind can by definition be called a thermal battery, as it has the ability to store heat. In the context of a house, that means dense materials like bricks, masonry and concrete.

Even a jug of water sitting in a sunny window is a thermal battery of sorts as it captures and later releases heat from the sun.

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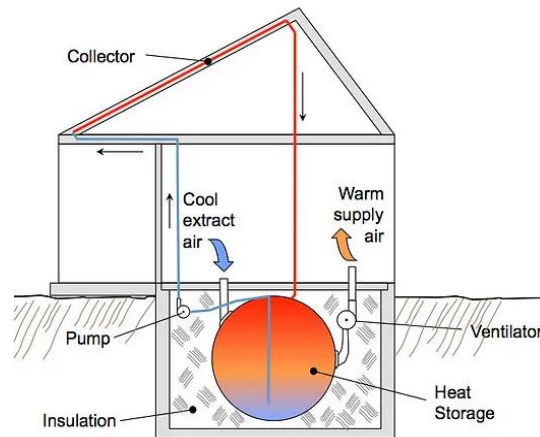


Heat Thermal Battery Storage

A well-insulated concrete floor acts as a thermal battery as well; once you pump it full of heat it takes a long time to cool down (depending on the thickness), and it regulates interior temperatures during that time.

One practical use for getting the most from a radiant concrete floor as a thermal battery would be in areas with fluctuating energy costs - you can set your floor on a timer so it only comes on during low-rate hours (7pm to 7am in Ontario for example).

During the twelve hours that it is off, it acts as a battery by slowly releasing the stored heat, so you avoid having to pay the higher rates during peak-hours.



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As you move into the area of active heat-storage systems, one of the more common types of thermal battery (not that there are a lot of them) is a huge water tank buried in the ground that is heated by solar thermal panels.

Even this type of system is not new, the first house in the United States with an active solar heating system was built In 1939 on the MIT campus (Massachusetts Institute of Technology), and sat on top of a huge water reservoir that was heated by thermal solar panels.



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What are phase change thermal batteries?

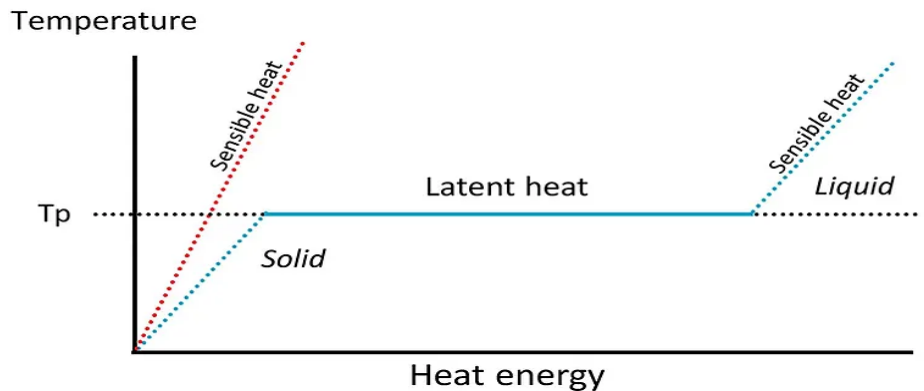
Taking advantage of a 'phase change' raises the bar a bit - stick with me, it will be fun, I promise :)

A significant input of energy is required to cause a material to change from a solid to liquid. That energy is later released when that material solidifies again.

While those transformations are happening and the material is either absorbing or releasing energy, the temperature will stay constant. Once the phase change is complete, the material will begin to change temperature again.

So what does that mean in real terms? It means that in order to melt water, wax, metal, rock or whatever, you need to feed it a ton of energy. but the temperature doesn't change while you are doing that. So your 'battery' has more power, and you can store more heat in the same volume of space.

It's difficult to take advantage of a melting point of 0° Celsius, but wax melts at about 37° Celsius (depending on its exact chemical makeup), which is perfect for collecting and storing heat from solar thermal collectors.



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How to build a thermal battery:

If you had a heat-collecting solar panel (directly heating air or liquid rather than generating power with photovoltaics), you can use that to charge your thermal battery. Envision this - a large tank of wax (or water) that is warmed by heated coils from a solar collector. Through that same tank runs another coil that is extracting the heat to pump it through your radiant floor or whatever other heating distribution system you have.

Specific Heat Capacity:

If you take solid paraffin (heat capacity $C_p = 2.5 \text{ kJ/kg}\cdot\text{K}$ and heat of fusion of 210 kJ/kg), let's say 1 kg , at room temperature, you will need 2.5 kJ (kilojoules) of heat to make the 1 kg block go from 20°C to 21°C . To make it go from 21°C to 22°C , you will also need 2.5 kJ (i.e. the same amount of energy).

Paraffin melts at approximately 37°C . If it drops to 36°C , you will again only need 2.5 kJ to bring it back to 37°C , but you would need 210 kJ (84 times as much) to go from 37 to 38°C .

This is because in order to melt, some chemical bonds in the solid lattice need to be broken, a process that requires extra energy. So overall, if a kg of paraffin is lying around at 20°C , you would need 252.5 kJ to bring it to 38°C .

One of the more common building materials with thermal mass benefits is concrete. In contrast to paraffin, 1 kg of concrete ($C_p = 0.88 \text{ kJ/kg}\cdot\text{K}$) would need 15.8 kJ to do the same. For water ($C_p = 4.18 \text{ kJ/kg}\cdot\text{K}$), the amount of energy required would be 75.2 kJ .

The amount of energy put in is the amount of energy stored in a material, as this energy will later be released as the material cools back down to 20°C , or room temperature. While there are many materials that can be used in the application of heat storage, this is just a quick comparison of some of the more commonly available ones.

So to conclude, paraffin can store 16 times as much heat per kg as concrete, and 3.4 times as much as water. So while water may not be the best material to store heat, it certainly is the most affordably priced and easily accessible.

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The Cp value referred to in the above text refers to the heat capacity of materials.

$$q = m C_p \Delta T$$

where:

q = energy [J]

m = mass of material [kg]

C_p = heat capacity of the material [kJ/(kg·K)]

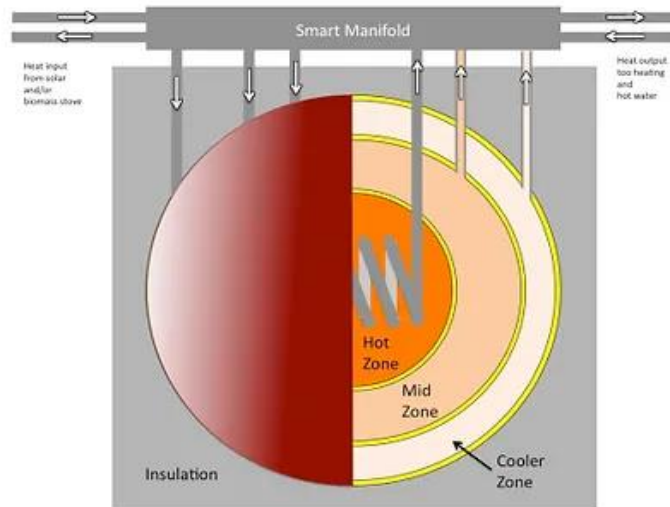
ΔT = temperature difference [K or °C]

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Heat Thermal Battery Storage

Hot Box Demo

This was one of the first demo models of the Hot Box. It uses sand (which has a very high specific heat) and a metal matrix to transfer the heat in and out. We could hold 33,000 BTU's storage per cubic foot...then a record!

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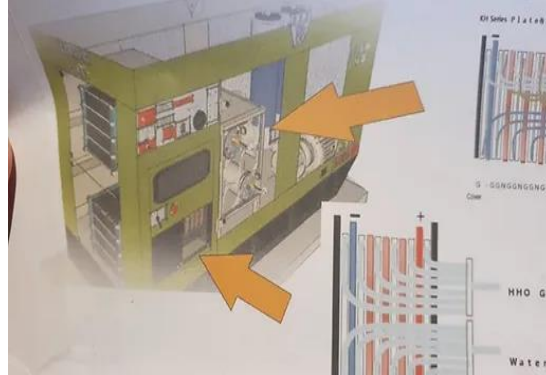
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Heat Thermal Battery Storage

Zeolite thermal storage retains heat indefinitely, absorbs four times more heat than water Secure Supplies Adds 1 Special Chemical to Unlock HIGH BTU with Long Electric Thermal Run Times.



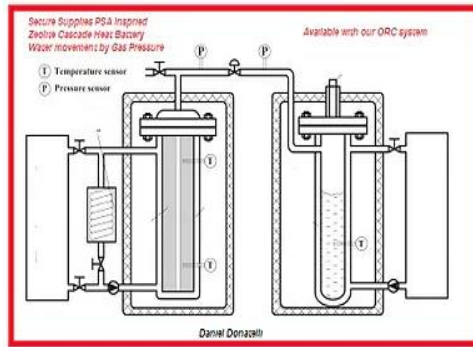
Secure Supplies

The Water to Heat Battery and Steam and Power Box is Ready for Ordering.

- Water to Gas
- Gas to Heat No Flame
- Heat to Heat Battery
- Heat to Power
- Heat to Steam for De Icing Work
- 10 x 5 x 4 Foot Fork Lift Skid
- Water Proof Rust Proof
- 24 Hr Run Times.

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Zeolite pellets that can store up to four times more heat than water, loss-free for “lengthy periods of time.” In theory, you can store heat in these pellets, and then extract exactly the same amount of heat after an indeterminate amount of time.

Zeolites (literally “boil stones”) aren’t exactly new: The term was coined in 1756 by Axel Cronstedt, a Swedish mineralogist who noted that some minerals, upon being heated, release large amounts of steam from water that had been previously adsorbed. For the last 250 years, scientists have tried to shoehorn this process in a heat storage system — and now, the Fraunhofer Institute, working with industrial partners, has worked out how to do it.

I will try to explain how this works, but the science is fairly complicated: When Fraunhofer’s zeolite comes into contact with water, a chemical reaction adsorbs the water and emits heat. When heat is applied to the zeolite, the process is reversed and the water is released. Because the heat is locked up in the chemical structure of the zeolite, the material never actually feels warm — which is why this is a “loss-free” storage method.

These two processes can be kept separate — so first you charge the balls up with heat, and then later you can just add water (!) to release the heat. This reaction occurs all along the surface of the zeolite — and because zeolites are porous, a single gram of the material has a surface area of 1000 square meters (10,700sqft). It is for this reason that Fraunhofer’s zeolite can store up to four times more heat than water.

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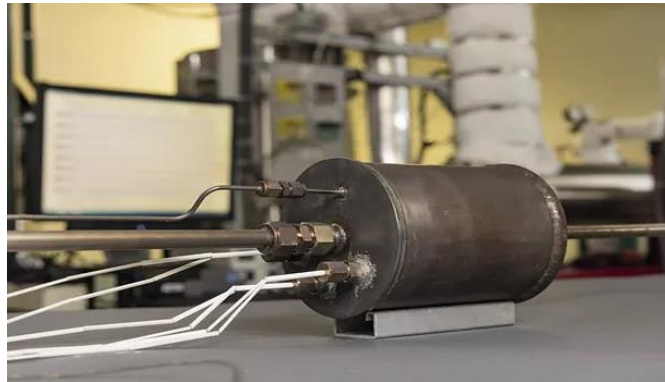
Large Commercial Hot Box

We had to build a unit that was pretty large and very cumbersome to store just over 2 million BTU's of heat. There was enough stored energy in one unit to make 2-3,000 gallons of hot water or heat a large building.
Improved Commercial Hot Box Model

We discovered a material that was 9 times the specific heat of sand. So, we could store 300,000 BTU's in one cu. ft. So, we could build the unit much smaller and wiser. This Hot Box is only 3' x 3' x 3' = 9 cu. ft. for 2 MBTU's+.-

Improved Commercial Hot Box Diagram

To get the heat out of the Hot Box, there is a thermostat in a closed mixing chamber that regulates the transfer of heat to keep the pressures in line. Air is less volatile than liquids to take the heat from the Hot Box below.
INSulated with Secure Supplies Special High Pressure Design At 1200 degrees C (over 2,000 degrees F)



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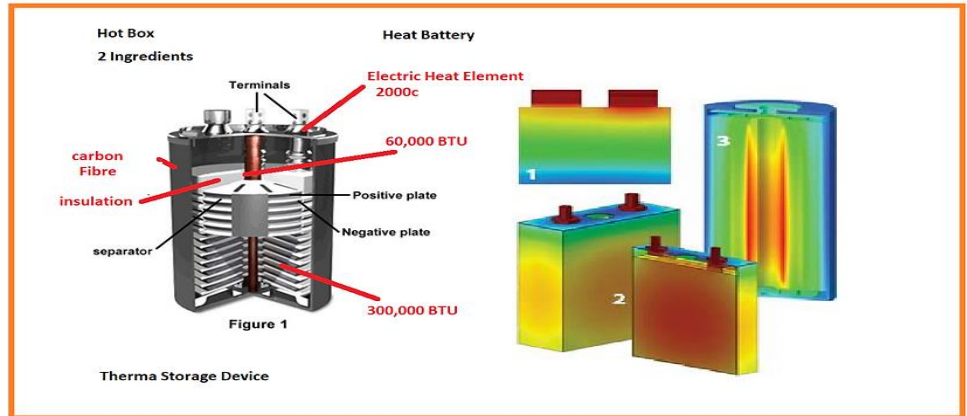
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Secure Supplies Example for Using Stored Heat

The Fischer Single or Double Stroke Engine This is a high torque, low RPM engine. It does not operate on the same RPM of the internal combustion engine.

Water Pressure No Room to turn to steam



Charge Battery Heat in From R 12A Or Catalytic or Electric Heat



8 Cylinder V8 Converted to 4 cylinder 2 stroke. with Water injection from battery Water Drained bottom of stroke, Crankcase is Sealed.

It has slower more deliberate power. For an automobile, it would require a change in transmission and drive train.

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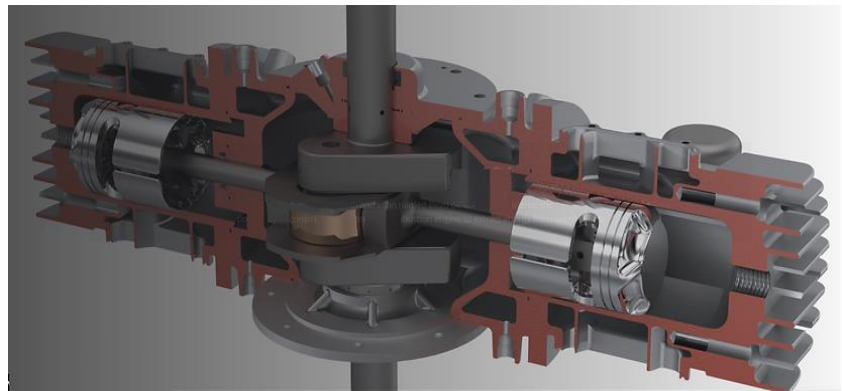
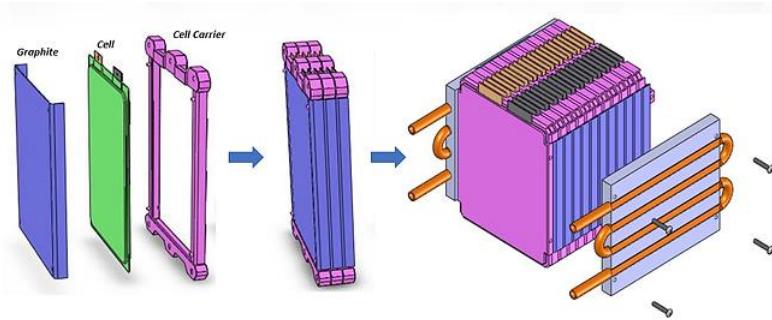
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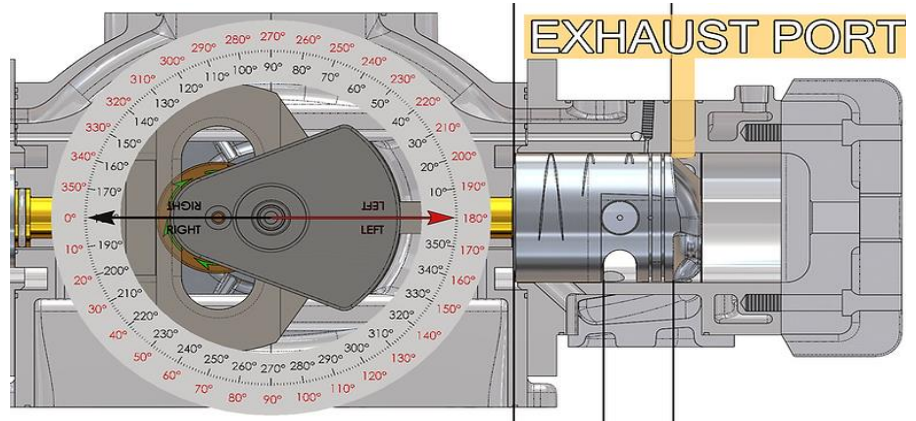
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Benefits:

- Mains pressure hot water
- Rapid reheat
- High discharge flow rate
- Compact Size
- Low heat losses, ERP A+
- No Pressure & Temperature safety valve
- No risk of explosion
- Negligible legionella risk
- No discharge pipework
- Flexible Siting, lower cost and easy install
- No need to comply with G3 regs*
- No maintenance requirements

Thermal Battery is Non Water Based Low Pressure Heat Pump For Domestic Hot Water and Heat Battery Storage homes.

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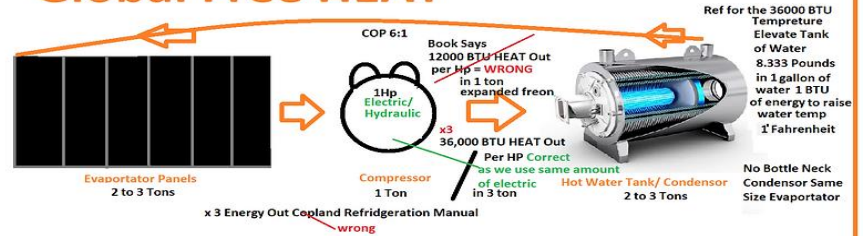
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Heat Thermal Battery Storage

Global Free HEAT



Refrigerant notes

Average Energy Coefficient of 6 to 1
 Carnot Cycle $\frac{200}{T1 - T2} = \frac{55cf}{200} = \% \text{ Electric Converted}$

Conversion of infinity Pump

3 Btu unit into Boiler Boiler 250 degree source 1 Electric bac to compressor 2 left over
 1 unit = 6 = 1/3 = 2 Electricity

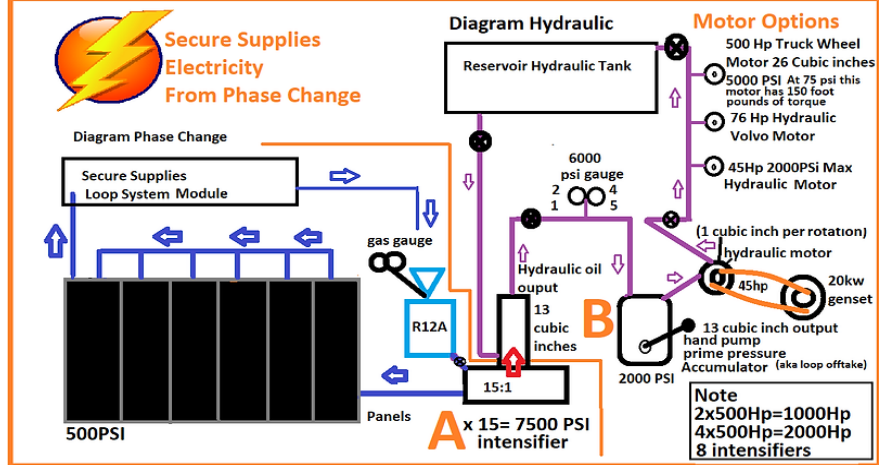


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Turbo Heat Fan runs off the heat of your woodstove.

The secret of the Free Heat fan is a small Stirling engine housed in the square base of the unit.

Stirling engines have been around for decades but no one has ever come up with a marketable way to use them.

They consist of a sealed chamber that expands quickly when heated and contracts when it cools down.

The heating up and cooling down happens almost instantly.

The piston in the small engine drives an offset crank that drives the fan.

The unit weighs about 15 lbs. and has a 14" diameter, 8-bladed fan. You just set the unit freely on top of a hot surface. Needs no batteries, fuel or electricity.

BASE PRICE
\$289.00

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